SHORT COMMUNICATION

Direct assessment of the antioxidant properties of midazolam by electron spin resonance spectroscopy

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Abstract Some antioxidant anesthetics directly inhibit lipid peroxidation mediated via the generation of reactive oxygen species (ROS). To date, the scavenging effects of midazolam on ROS have not been directly assessed. We investigated the inhibitory effect of midazolam on ROS [hydroxyl radical (HO') and superoxide (O_2^-)] by in vitro X-band electron spin resonance with the spin-trapping agent 5,5-dimethyl-1-pyrroline-*N*-oxide. Our results indicated that HO' and O_2^- were not affected by midazolam at clinically relevant concentrations, but were directly scavenged by midazolam at high concentrations (i.e., >4.6 and >1.5 mM, respectively).

Keywords Reactive oxygen species · Oxidative stress · Electron spin resonance (ESR) · Antioxidants · Midazolam

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Division of Anesthesiology, Department of Clinical Care Medicine, Kanagawa Dental College, Yokosuka, Kanagawa, Japan Various reports have shown that oxidative stress induced by reactive oxygen species (ROS) such as the hydroxyl radical (HO) and superoxide (O_2^-) can contribute to the pathophysiology of brain diseases, including ischemia– reperfusion injury [1], tumor [2], aging [3], and various neurodegenerative disorders [4]. The protective effects of general anesthetics against oxidative stress are well known [5–7]. Antioxidant anesthetics, particularly propofol (2,6-diisopropylphenol), have been shown to inhibit ROSmediated lipid peroxidation [5, 6, 8]. In a previous study using electron spin resonance (ESR), we demonstrated that propofol with a carrier emulsion consisting of mediumchain triglycerides (MCT) and long-chain triglycerides (LCT) acted as a scavenger of HO in the brains of strokeprone spontaneously hypertensive rats (SHRSPs) [9].

Midazolam and propofol are commonly used as sedatives for critically ill patients [10]. Such patients usually suffer from the pathologic effects of oxidative stress, defined as an imbalance between the generation of ROS and inactivation by antioxidant defense systems. Therefore, the antioxidant activities of propofol and midazolam may be of clinical importance. Propofol pretreatment greatly attenuated the impairment in comparison with midazolam, which agrees with the concept of antioxidant activity in the presence of membranes [10]. Propofol has previously been shown to have a greater antioxidant effect than midazolam against peroxyl radicals generated in the presence of erythrocyte membranes [10]. However, to date, the direct scavenging effects of midazolam on ROS have not been examined, for example by using a direct detection method such as ESR.

Our laboratories have been developing in vitro biomedical applications for the ESR detection of ROS such as O_2^- [11], H_2O_2 [12], 1O_2 [13], and HO⁻ [14]. We have successfully applied this method to assess the antioxidant properties of drugs in biological systems [9]. In the present study, we used the same technique to investigate the antioxidant effects of midazolam. We found that midazolam scavenged ROS at high concentrations, but that the antioxidant effect of midazolam was less potent than that of propofol.

Midazolam was purchased from Astellas Pharma Inc. (Tokyo, Japan). Xanthine oxidase [XO; (grade III: from bovine milk, chromatographically-purified suspension in 2.3 M (NH_4)₂SO₄–10 mM sodium phosphate buffer, pH 7.8, containing 1 mM EDTA and 1 mM sodium salicylate)], xanthine, and superoxide dismutase were obtained from Sigma (St. Louis, MO, USA). Hydrogen peroxide (H_2O_2) and FeSO₄ were obtained from Wako Chemical (Osaka, Japan). 5,5-Dimethyl-1-pyrroline-*N*-oxide (DMPO) was purchased from Labotec (Tokyo, Japan).

HO was generated by the Fenton reaction (H_2O_2 plus FeSO₄) and by ultraviolet (UV) irradiation of H_2O_2 , as described previously [9]. O_2^- was generated from xanthine–XO by methods also described previously [9]. All solutions were prepared in 0.1 M phosphate-buffered saline (pH 7.2). ESR spin-trapping was conducted with an ROS generating system containing DMPO. ESR observations were performed with a JES-RE 1X X-band spectrometer (JEOL, Tokyo, Japan) connected to a WIN-RAD ESR Data Analyzer (Radical Research, Tokyo, Japan) at the following instrument settings: microwave power 8.00 mW, magnetic

field 335.0 ± 5 mT, field modulation width 0.079 mT, receiver gain 400–500, sweep time 1 min, and time constant 0.03 s. To quantify the spin adducts detected, we obtained ESR spectra for manganese oxide standards. After the ESR spectra had been recorded, the signal intensity (expressed as relative height) was normalized against the signal intensity of the manganese oxide standard marker [9]. All experiments were repeated a minimum of three times. The scavenging effect was considered to correspond to the signal intensity (percent of control), as described previously [15, 16].

Statistical analyses were performed using Ekuseru-Toukei 2008 (Social Survey Research Information Co., Ltd., Tokyo, Japan). Results are presented as the mean \pm SD. The Dunnett test was used for multiple comparisons. A *p* value of less than 0.05 was considered to be statistically significant.

In the present study, we used ESR spin-trapping with DMPO to assess the effect of midazolam on HO generated by the Fenton reaction. As reported previously [9], following the addition of H_2O_2 to FeSO₄ a characteristic DMPO–OH spin adduct with hyperfine splitting giving rise to four resolved peaks was observed (Fig. 1a; control). Midazolam at high concentrations (7.7–10 mM) reduced the DMPO–OH signal in a dose-dependent manner (Fig. 1).

In the next study, ESR spin-trapping with DMPO was used to assess the effect of midazolam on HO generated by





b

Fig. 1 Scavenging effects of midazolam on hydroxyl radicals (HO') generated from the Fenton reaction. **a** Electron spin resonance (ESR) spin-trapping measurement of HO' generated from H_2O_2 (20 μ M) and FeSO₄ (20 μ M) in 0.1 M phosphate-buffered saline (pH 7.2) with the spin trap 5,5-dimethyl-1-pyrroline-*N*-oxide (50 mM) in the absence of midazolam (control) and with midazolam pretreatment at 6.1, 7.7, 9.2, and 10.0 mM. Signals appearing on either side of the ESR spectrum

correspond to manganese oxide (MnO), used as a reference. **b** Dose– response showing the scavenging effect of midazolam (*closed column* 6.1–10.0 mM) on HO generated from the Fenton reaction. The signal intensity of the second peak of the spectrum was normalized to the signal intensity of the MnO standard. Data represent mean \pm SD of three identical experiments. Values annotated with *asterisks* were significantly different (p < 0.05) from the corresponding control value





b

Fig. 2 Scavenging effects of midazolam on hydroxyl radicals (HO) generated from the ultraviolet (UV) irradiation of H_2O_2 . **a** Electron spin resonance (ESR) spin-trapping measurement of HO generated from the UV irradiation (365 nm, 40 mW) of H_2O_2 (20 mM) in 0.1 M phosphate-buffered saline (pH 7.2) with the spin trap 5,5-dimethyl-1-pyrroline-*N*-oxide (50 mM) in the absence of midazolam (control) and with midazolam pretreatment at 1.5, 4.6, 6.1, and 10.0 mM. Signals appearing on either side of the ESR spectrum

UV irradiation of H_2O_2 . Consistent with previous studies [14, 17, 18], a characteristic DMPO–OH spin adduct spectrum with hyperfine splitting giving rise to four resolved peaks (Fig. 2a; control) was observed following the UV irradiation of H_2O_2 . The DMPO–OH spin adduct was significantly reduced at high concentrations (>4.6 mM) (Fig. 2a). Pretreatment with midazolam (4.6–10.0 mM) reduced the DMPO–OH signal in a dose-dependent manner (Fig. 2b). These data suggest that high concentrations of midazolam directly scavenged HO by a mechanism that did not involve iron chelation (Fig. 1).

We also investigated the scavenging effects of midazolam on O_2^- , as measured by ESR spin-trapping with DMPO, in the presence of midazolam. As reported previously [9, 18], after the addition of xanthine to XO, a characteristic DMPO–OOH spin adduct with hyperfine splitting giving rise to 12 resolved peaks was observed (Fig. 3a; control). The addition of high concentrations of midazolam (1.5–6.1 mM) to the xanthine–XO system led to a concentration-dependent decrease in the intensity of the DMPO–OOH signal (Fig. 3).

When given intravenously for sedation, midazolam has been shown to decrease ROS production under conditions of surgical stress [19]. The potential neuroprotective effect of intravenous anesthetics such as propofol may be mediated by their antioxidant properties, which have been shown to inhibit apoptosis, ischemia–reperfusion injury,

correspond to manganese oxide (MnO), used as a reference. **b** Dose– response showing the scavenging effect of midazolam (*closed column* 1.5–10.0 mM) on HO generated from the UV irradiation of H₂O₂. The signal intensity of the second peak of the spectrum was normalized to the signal intensity of the MnO standard. Data represent mean \pm SD of three identical experiments. Values annotated with *asterisks* were significantly different (p < 0.05) from the corresponding control value

and inflammatory-induced neuronal damage [20, 21]. Several prior studies have suggested that midazolam does not have antioxidant effects, based on its inability to inhibit lipid peroxidation [22, 23]. Previous studies have shown that midazolam can inhibit lipid peroxidation in vitro [24]; however, it was not clear whether this was due to a direct scavenging effect on ROS. We therefore undertook the present ESR study to directly assess the effect of midazolam on HO and O_2^- . Our results provide direct evidence that high concentrations of midazolam scavenge HO as well as O_2^- generated from xanthine–XO (Figs. 1, 2, 3).

First, we showed that midazolam dose-dependently scavenged HO generated by the Fenton reaction (Fig. 1). This type of iron-dependent generation of HO has been implicated in the pathophysiologies of various disease states; for example, DMPO–OH spin adducts can be detected in the synovial fluid of patients with temporo-mandibular disease [25], and in gingival fibroblasts from patients with Down's syndrome [26].

We also investigated the effect of midazolam on HO generated by the UV irradiation of H_2O_2 , which is known to be an Fe²⁺-independent reaction [14, 17]. Interestingly, high concentrations of midazolam dose-dependently inhibited the HO generated via the UV irradiation of H_2O_2 (Fig. 2). Propofol MCT/LCT was previously shown to scavenge HO in a dose-dependent manner in this system [9], and the inhibitory effects of propofol MCT/LCT were observed at

6.1 mM 330 332 334 336 338 340 magnetic field (mT) **Fig. 3** Scavenging effects of midazolam on superoxide (O_2^{-})

а

control



generated by the addition of xanthine to XO. a Electron spin resonance (ESR) spin-trapping measurement of O_2^{-} generated from xanthine oxidase (XO; 0.1 U/ml) and xanthine (362 µM) in 0.1 M phosphate-buffered saline (pH 7.2) with the spin trap 5,5-dimethyl-1-pyrroline-N-oxide (440 mM) in the absence of midazolam (control) and with midazolam pretreatment at 0.15, 1.5, 3.0, and 6.1 mM. Signals appearing on either side of the ESR spectrum correspond to

concentrations lower than those required with midazolam in the Fenton system (>7.7 mM; Fig. 1) and H₂O₂/UV system (>4.6 mM; Fig. 2). Midazolam at high concentrations (>1.5 mM) was also found to be a scavenger of O_2^- generated by xanthine–XO (Fig. 3), but it did not scavenge O_2^{-1} treated with the propofol MCT/LCT (original liquid 5.609 mM) [9].

These results indicate that midazolam and propofol MCT/ LCT differ in their ROS-scavenging properties and that this difference may be clinically significant. In the case of propofol, antioxidant activity seems to be mediated by hindered phenolic structures [27], as in the case of butylated hydroxytoluene, butylated hydroxyanisole, and tocopherols [28], whereas for midazolam, the structure responsible is unclear. However, in a previous study on stobadine, a pyridoindole derivative that possesses potent antioxidant activity, Kagan et al. [29] noted the possibility that intrinsic nitrogen atoms might act as an antioxidant by donating electrons to free radicals. This might also be the case with midazolam.

The lowest concentration of midazolam used clinically was about 0.3 µM; the minimum concentrations required for inhibition ranged from 1.5 mM for O₂⁻⁻ to 4.6–7.7 mM for HO. This finding contrasts with our previous ESR study, which showed ROS-scavenging effects of propofol MCT/LCT at concentrations below 112 μ M [9]. These

manganese oxide (MnO), used as a reference. b Dose-response showing the scavenging effect of midazolam (closed column 0.15–6.1 mM) on O_2^- generated from XO and xanthine. The signal intensity of the second peak of the spectrum was normalized to the signal intensity of the MnO standard. Data represent mean \pm SD of three identical experiments. Values annotated with asterisks were significantly different (p < 0.05) from the corresponding control value

results suggest that the antioxidant activity of propofol MCT/LCT may be therapeutically beneficial in patients with diseases associated with ROS. Propofol was also shown to be more effective than midazolam at reducing oxidative stress in a homogeneous solution containing erythrocyte membranes [24]. Interestingly, O_2^- scavenging was observed with high concentrations of midazolam in our study (Fig. 3), but not with propofol MCT/LCT in a previous study [9]. This finding may be clinically relevant because O₂⁻⁻ has been shown to play a role in ischemiareperfusion injury [30].

In conclusion, this in vitro ESR study showed that the O_2^{-} and HO^{\cdot} can be scavenged by midazolam, but this effect occurs only at high rather than clinically relevant concentrations.

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